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Problems Problem 1.1 Write
and compare the ground-state
electron configurations for
each pair of elements: (a)
Carbon and silicon C (6

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Solutions) $1s^2 2s^2 2p^2$ Si
(14 electrons) $1s^2 2s^2 2p^6 3s^2 3p^2$ Both carbon and
silicon have four electrons
in their outermost (valence)
shells.

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William H. Brown is emeritus professor of chemistry at Beloit College, where he was twice named Teacher of the Year. His teaching responsibilities include

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Solutions organic chemistry, advanced organic chemistry, and, more recently, special topics in pharmacology and drug synthesis.

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